Chinese Chemical Society | Xiamen University Journal of Electrochemistry

Online First

9-25-2023

Micropatterning and Functionalization of Single Layer Graphene: Tuning Its Electron Transport Properties

Miaomiao Cui Lianhuan Han Lanping Zeng Jiayao Guo Weiying Song Chuan Liu Yuanfei Wu Shiyi Luo Yunhua Liu Dongping Zhan

DOI: 10.13208/j.electrochem.2305251

2 <u>Http://electrochem.xmu.edu.cn</u>



Micropatterning and Functionalization of Single Layer Graphene: Tuning Its Electron Transport Properties

Miaomiao Cui,^a Lianhuan Han,^{a,b*} Lanping Zeng,^a Jiayao Guo,^a Weiying Song,^a Chuan Liu,^a Yuanfei
Wu,^a Shiyi Luo,^a Yunhua Liu,^{c*} Dongping Zhan^{a*}

7aDepartment of Chemistry, College of Chemistry and Chemical Engineering; State Key Laboratory of Physical Chemistry of Solid Surfaces (PCOSS);88Engineering Research Center of Electrochemical Technologies of Ministry of Education; Xiamen University; Xiamen 361005, China.99E-mail: dpzhan@xmu.edu.cn

- 10bDepartment of Mechanical and Electrical Engineering, Pen-Tung Sah Institute of Micro-Nano Science and Technology, Xiamen University; Xiamen 361005,11China.
- 12 E-mail: hanlianhuan@xmu.edu.cn
- 13 c National CAD Support Software Engineering Research Center Huazhong University of Science and Technology, Wuhan 430074, China.
- 14 E-mail: liuyh@mail.hust.edu.cn
- 15

1

16 Keywords: Graphene patterning, Electron transport, Electrochemical bromination, Photolithography, All graphene device

17 Abstract: As a promising 2D material, graphene exhibits excellent 18 physical properties including single-atom-scale thickness and 19 remarkably high charge carrier mobility. However, its semi-metallic 20 nature with a zero bandgap poses challenges for its application in 21 high-performance field-effect transistors (FETs). In order to 22 overcome these limitations, various approaches have been explored 23 to modulate graphene's bandgap, including nanoscale confinement, 24 external field induction, doping, and chemical micropatterning. 25 Nevertheless, the stability and controllability still need to be 26 improved. In this study, we propose a feasible method that combines 27 electrochemical bromination and photolithography to precisely tune 28 the electron transport properties of single layer graphene (SLG). 29 Through this method, we successfully fabricated various brominated 30 SLG (SLGBr) micropatterns with high accuracy. Futher investigation 31 revealed that the electron transport properties of SLG can be 32 conveniently tuned by controlling the degree of bromination. The 33 SLGBr exhibited a resistance, and have a decreasing conductance 34 with the bromination degree increasing. When the bromination 35 degree increased to a critical value, the SLGBr demonstrated 36 semiconducting characteristics. This research offers a prospective 37 route for the fabrication of graphene-based devices, providing 38 potential applications in the realm of microelectronics.

39 Graphene has been considered as the most 40 prospective material because of its excellent properties, 41 including single-atom-scale thickness^[1], extremely high charge carrier mobility^[2], superior optical transparency^[3], 42 43 high thermal conductivity^[4], etc.. Recently, the mass 44 production of high-quality single layer graphene (SLG) at 45 wafer scale makes it qualified in various possible 46 applications such as micro/nano-electronics, sensors and 47 actuators, microelectromechanical system (MEMS), etc.^[5-7]. 48 To this, the localized micropatterning and functionalization 49 of SLG at wafer scale is crucial in order to tune the physical 50 and chemical properties, e.g., band gap, electron 51 conductivity, transmittance, wettability, etc., and to fulfil the 52 requirements of various industrial domains^[8].

Great efforts have been made to fabricate SLG patterns. Physical methods can be classified into top-down and bottom-up categories^[9]. The top-down methods are to remove the carbon atoms from pristine SLG wafer by the energy-beam techniques including the focused laser beam, electron beam, plasma beam, reactive ion beam, etc.^[10]. The top-down methods can be realized by either direct writing or template forming combined with various lithography techniques. The bottom-up methods are to synthesize specially sized SLG on the patterned catalysts such as copper, nickel, platinum, gold, etc.[11-13]. The most advantage of physical methods lies that the crystalline structure of SLG are not destroyed. Consequently, the intrinsic properties of SLG are kept very well. Nevertheless, the physical properties of SLG, e.g., the band gap, can be regulated by the external physical field^[14]. However, when the external physical field were dismissed, the band gap of SLG would go back to zero.

Comparing physic methods, chemical micropatterning can change the crystalline structure and thus, change the physical and chemical properties of SLG^[8]. Chemical adsorption^[15] or doping^[16] have been adopted to tune the SLG's properties by introducing the sp³-C defects. However, after thermal annealing, many of the functionalized SLG materials change back to be pristine due to the weak interaction between SLG and the adsorbates or dopants, e.g., van der Waals' force^[17], adsorption bond^[18], or even chemical bond^[19]. To this, we developed an in-situ electrochemically induced radical addition reaction to functionalize SLG with high efficiency and stability^[20, 21]. Here we report that the functionalized SLG micropatterns has tuneable electron transport properties, making SLG act as single-atomic-thickness resistor or rectifier, depending on the introduced sp³-C defects density.

The electrochemical micropatterning and functionalization procedures of SLG are depicted as in Scheme 1. First, a thin-layer viscosifier (HMDS) was spin-





Scheme 1 Preparing processes of brominated SLG with patterns. (i) photoresist coating, (ii) Photolithography, (iii) electrochemical brominating.

1 coated on the SLG wafer to increase the adhesion between 2 SLG and photoresist. Second, a thin layer of photoresist 3 (AZ5214E) film was spin-coated on the substrate. Then, the 4 photoresist film was exposed by the laser-beam direct-5 writing photolithography. Later on, the exposed photoresist was removed by tetramethylammonium hydroxide for 6 7 several times and the aimed SLG micropattern array was 8 obtained. After that, the obtained SLG micropattern array 9 was immersed in an aqueous electrolyte containing 10 mM 10 KBr and 10 mM H₂SO₄ to performed bromination addition 11 reaction by cyclic voltammetry in a potential range from -12 0.75 to 1.2 V vs. Hg/Hg₂SO₄ reference electrode with a 13 scan rate of 100 mV/s for certain cycles (See Figure S1). 14 Finally, the brominated SLG (SLGBr) micropatterns were 15 obtained after removing the photoresist protection layer by 16 rinsing with acetone and isopropanol.

17 As reported very recently by us, the functionalization 18 of SLG is realized by the bromination addition reaction^[20]. 19 Because SLG is conductive enough to act as the working 20 electrode, bromide anions are oxidized to bromine radicals, 21 which can react with SLG to form SLGBr. When SLG 22 changes into SLGBr, the interfacial electrochemical reaction 23 activity will be decreased because the Faraday current 24 keeping decreasing with the potential scanning going on. The SLGBr micropatterns "XMU", the acronym of Xiamen 25 26 University, were observed to have a different colour with 27 SLG by optical microscopy (Figure 1a). The corresponding images of scanning electron microscopy (SEM) and 28 29 scanning Raman microscopy (SRM) were shown in Figure 30 1b and 1c. The clear boundaries between the SLG and 31 SLGBr were also observed, indicating the different 32 properties between them.

The Raman spectrum of the SLGBr region is much different from that of the SLG region (Figure 1d). The pristine SLG has a G-peak at near 1586 cm⁻¹ and a 2D peak at 2687 cm⁻¹. The 2Dpeak is sharp and symmetrical, and the peak intensity is about 4 times larger than that of G-peak, indicating that the pristine SLG has a perfect lattice structure^[22]. After the functionalization by bromination addition reaction, the newly observed D-peak (~1350 cm⁻¹) and D'-peak (~1620 cm⁻¹) represents the sp³-C defects^[23], indicating that the symmetry of the lattice is changed due to the formation of C-Br bond. Furthermore, the G-peak blue-shifted to 1597 cm⁻¹ at the brominated region, and 2D peak blue-shifted to 2692 cm⁻¹, predicting that SLGBr is p-type doping^[24], which is consistent with the strong electron absorption ability of bromine.





Figure 1 The optical image (a) and SEM image (b) of patterned SLGBr with the "XMU" logo.(c) Raman I_D/I_G mapping of the patterned SLGBr. (d) The corresponding Raman spectra of pristine SLG and SLGBr. (e) Statistical I_D/I_G histograms from the mapping image of (c).



1 The intensity ratio (I_D/I_G) is usually adopted to 2 characterize the defect density of graphene, that is, the 3 ratio of the sp³ to sp² bonding character. A higher the I_D/I_G 4 ratio means a higher defect density^[25]. The statistical 5 analysis of SRM image gives a distribution of ID/IG ratio as 6 shown in Figure 1c. The I_D/I_G ratio in the "XMU" region is 7 much higher than that in the remaining region, which 8 indicates that photoresist micropattern can protect well the 9 underlying graphene and let the bromination addition 10 reaction selectively occur in the exposed area. The 11 statistical I_D/I_G-ratio histogram was employed to analyzed 12 the bromination degree of SLG. From Figure 1e, I_D/I_G 13 ranged from 0.5 to 1.5 with a mid-value about 1.2, 14 suggesting that the bromination degree of SLGBr 15 micropattern was pretty uniform.

16 To demonstrate the versatility and reliability of the 17 method, SLGBr micropattern array such as square ring, 18 cross and an H- type Chinese knot were fabricated. The 19 optical microscopy images of the SLG micropatterns after 20 laser-beam direct-writing photolithography were shown in 21 Figure 2a-2c, and the relevant optical images of the SLGBr 22 micropatterns were shown in Figure. S2, indicating the clear 23 SLG/SLGBr boundaries. The relevant ID/IG distribution of 24 the SLGBr micropatterned SLG was characterized by the 25 SRM as shown in Figure 2d-2f, indicating the uniform 26 functionalization of bromination addition reaction.

27 We have demonstrated that the electron transport (i.e., 28 electronic properties) and interfacial electron transfer (i.e., 29 electrochemical properties) of functionalized SLG will change a lot from the pristine SLG^[20]. To investigate the electron transport 30 31 properties of SLGBr, we designed and fabricated the SLG-32 SLGBr interdigitated electrodes with a pair of Cr/Au belts as the 33 current collectors (Scheme 2) to construct a field-effect transistor 34 (FET), where the SLG electrode acts as the drain, the SLGBr 35 acts as the source, and the Si/SiO₂ substrate acts as the gate. In 36 order to investigate the characteristics of the devices, SLG



Figure 3 (a-c) The I_{DS} - V_{DS} curves of the SLGBr Interdigitated Electrode when I_D/I_G value is 0.5, 1.7 and 2.4, respectively. Inset: the slope of the I_{DS} - V_{DS} curves changes with V_{GS} . (d) The statistic relationship curve between the bromination degree (I_D/I_G) and the conductance (dI_{DS}/dV_{DS}) at V_{GS} =0 V.

electrodes and SLGBr electrodes were prepared simultaneously. The SLG electrodes exhibit resistance characteristics and p doping due to the residual photoresist (Figure S 3a-3b), while the SLGBr electrodes with higher defect density showed semiconducting characteristics (Figure S 3c-3d). Akin to chemical functionalization, electrochemically induced bromination addition introduced new sp³ C-Br bond to SLG. Because of the asymmetric broken honeycomb structure, the bandgap of SLGBr rises, and a semimetal–semiconductor transition takes place in SLGBr.



Figure 2 (a-c) Optical images of SLG with photoresist patterns, including (a square rings, (b) crosses and (c) H- type Chinese knot. (d-f) Corresponding Raman I_D/I_G mapping of the SLGBr patterns.



Scheme 2 (a-g) Preparing processes of the graphene Interdigitated Electrodes with designed SLGBr patterns. (h) The optical image of a typical Interdigitated Electrode with patterned SLGBr.

1 Figure 3a-3c show the IDS-VDS curves of the SLGBr/SLG FET 2 with different bromination degree characterized by the statistical 3 I_D/I_G mid-values from the SRM images of SLGBr. With the low 4 bromination degree (I_D/I_G : 0.5 in Figure 3a), the I_{DS} - V_{DS} curves 5 shows a typical electronic property of a resistor. The 6 conductance, i.e., the slope of the IDS-VDS curves changes 7 linearly with V_{GS} in a potential region [-50 V, +50 V], shown as 8 the insert of Figure 3a. The voltage-sensitive feature 9 demonstrates the potential application of SLG into micro-varistor. 10 With the higher bromination degree (I_D/I_G : 1.7 in Figure 3b or 2.4 11 in Figure 3c), the I_{DS}-V_{DS} curves were obviously different from 12 those in Figure 3a, presenting the characteristics of current 13 rectification. The results show that the SLGBr will be changed 14 into a semiconductor if the bromination degree becomes 15 higher^[20, 26]. This result is similar to our previous study, but due to the low degree of bromination, the semiconductor 16 17 characteristics are less significant than before^[23]. We performed 18 the electronic measurements of the FETs with different 19 bromination degree and different V_{GS} bias, and the statistic 20 relationship between the conductance (dl_{DS}/dV_{DS}) and the 21 bromination degree (I_D/I_G) is shown in Figure 3d. The critical I_D/I_G 22 value between conductor and semiconductor can be estimated 23 as 0.6. The mechanisms of the change of electron transport 24 properties needs to be further studied in future.

25 In conclusion, laser beam direct-writing photolithography and 26 electrochemically induced bromination addition reaction were 27 adopted for the micropatterning and functionalization of SLG. 28 We found that the brominating functionalization can tune the 29 electron transport properties of SLG by controlling its bromination degree. The conductance of SLGBr keeps 30 31 decreasing at low bromination degree. After a critical value, 32 SLGBr will become a semiconductor. This work demonstrates 33 the possibility for the fabrication of ultrathin microelectronic 34 devices by SLG.

35 Supporting Information

36 Supporting information for this article is given via a link at the 37 end of the document.((Please delete this text if not appropriate.))

The financial support by the National Natural Science Foundation of China (21827802, 22202166, 22132003, 22021001) and the 111 Project (B08027, B17027) are appreciated. We thank Tan Kah Kee Innovation Laboratory for providing characterization services.

References

- Kaplan A, Yuan Z, Benck JD, Govind Rajan A, Chu XS, Wang QH, Strano MS. Current and Future Directions in Electron Transfer Chemistry of Graphene[J]. Chem. Soc. Rev.,2017, 46(15): 4530-4571.
- [2] Morozov SV, Novoselov KS, Katsnelson MI, Schedin F, Elias DC, Jaszczak JA, Geim AK. Giant Intrinsic Carrier Mobilities in Graphene and Its Bilayer[J]. Phys. Rev. Lett., 2008, 100(1): 016602.
- [3] Nair RR, Blake P, Grigorenko AN, Novoselov KS, Booth TJ, Stauber T, Peres NM, Geim AK. Fine Structure Constant Defines Visual Transparency of Graphene[J]. Science, 2008, 320(5881): 1308.
- [4] Nan HY, Ni ZH, Wang J, Zafar Z, Shi ZX, Wang YY. The Thermal Stability of Graphene in Air Investigated by Raman Spectroscopy[J]. Journal of Raman Spectroscopy,2013, 44(7): 1018-1021.
- [5] He Q, Wu S, Yin Z, Zhang H. Graphene-Based Electronic Sensors[J]. Chem. Sci., 2012, 3(6): 1764–1772.
- [6] Biro LP, Nemes-Incze P, Lambin P. Graphene: Nanoscale Processing and Recent Applications[J]. Nanoscale,2012, 4(6): 1824-1839.
- [7] Schwierz F, Pezoldt J, Granzner R. Two-Dimensional Materials and Their Prospects in Transistor Electronics[J]. Nanoscale,2015, 7(18): 8261-8283.
- [8] Wei T, Bao L, Hauke F, Hirsch A. Recent Advances in Graphene Patterning[J]. Chempluschem, 2020, 85(8): 1655-1668.
- [9] Wei T, Hauke F, Hirsch A. Evolution of Graphene Patterning: From Dimension Regulation to Molecular Engineering[J]. Adv. Mater., 2021, 33(45): 2104060.
- [10] Zheng YQ, Wang H, Hou SF, Xia DY. Lithographically Defined Graphene Patterns[J]. Advanced Materials Technologies, 2017, 2(5): 1600237
- [11] Park JU, Nam S, Lee MS, Lieber CM. Synthesis of Monolithic Graphene-Graphite Integrated Electronics[J]. Nat. Mater., 2011, 11(2): 120-125.
- [12] Choi J-K, Kwak J, Park S-D, Yun HD, Kim S-Y, Jung M, Kim SY, Park K, Kang S, Kim S-D, Park D-Y, Lee D-S, Hong S-K, Shin H-J, Kwon S-Y. Growth of Wrinkle-Free Graphene on Texture-Controlled Platinum Films and Thermal-Assisted Transfer of Large-Scale Patterned Graphene[J]. ACS Nano,2015, 9(1): 679-686.
- [13] Zhou X, Qi Y, Shi J, Niu J, Liu M, Zhang G, Li Q, Zhang Z, Hong M, Ji Q, Zhang Y, Liu Z, Wu X, Zhang Y. Modulating the Electronic Properties of

38 Acknowledgements

- 1 Monolayer Graphene Using a Periodic Quasi-One-Dimensional Potential 2 Generated by Hex-Reconstructed Au(001)[J]. ACS Nano,2016, 10(8): 3 7550-7557. 4 [14] Zhang Y, Tang TT, Girit C, Hao Z, Martin MC, Zettl A, Crommie MF, Shen 5 YR, Wang F. Direct Observation of a Widely Tunable Bandgap in Bilayer 6 Graphene[J]. Nature, 2009, 459(7248): 820-823. 7 [15] Balog R, Jørgensen B, Nilsson L, Andersen M, Rienks E, Bianchi M, 8 Fanetti M, Lægsgaard E, Baraldi A, Lizzit S, Sljivancanin Z, Besenbacher F, 9 Hammer B, Pedersen TG, Hofmann P, Hornekær L. Bandgap Opening in 10 Graphene Induced by Patterned Hydrogen Adsorption[J]. Nat. Mater., 2010, 11 9(4): 315-319. 12 [16] Wu J, Xie L, Li Y, Wang H, Ouyang Y, Guo J, Dai H. Controlled Chlorine 13 Plasma Reaction for Noninvasive Graphene Doping[J]. J. Am. Chem. 14 Soc.,2011, 133(49): 19668-19671. 15 [17] Yavari F, Kritzinger C, Gaire C, Song L, Gulapalli H, Borca-Tasciuc T, 16 Ajayan PM, Koratkar N. Tunable Bandgap in Graphene by the Controlled 17 Adsorption of Water Molecules[J]. Small,2010, 6(22): 2535-2538. 18 [18] Elias DC, Nair RR, Mohiuddin TM, Morozov SV, Blake P, Halsall MP, 19 Ferrari AC, Boukhvalov DW, Katsnelson MI, Geim AK, Novoselov KS. 20 Control of Graphene's Properties by Reversible Hydrogenation: Evidence 21 for Graphane[J]. Science, 2009, 323(5914): 610-613. 22 [19] Wei T, Kohring M, Chen M, Yang S, Weber HB, Hauke F, Hirsch A. Highly 23 Efficient and Reversible Covalent Patterning of Graphene: 2d-Management 24 of Chemical Information[J]. Angew. Chem. Int. Ed. Engl., 2020, 59(14): 25 5602-5606 26 [20] Zeng L, Song W, Jin X, He Q, Han L, Wu Y-f, Lagrost C, Leroux Y, Hapiot 27 P, Cao Y, Cheng J, Zhan D. Electrochemical Regulation of the Band Gap 28 of Single Layer Graphene: From Semimetal to Semiconductor[J]. Chem. 29 Sci..2023 30 [21] Chen DH, Lin Z, Sartin MM, Huang TX, Liu J, Zhang QG, Han LH, Li JF, 31 Tian ZQ, Zhan DP. Photosynergetic Electrochemical Synthesis of 32 Graphene Oxide[J]. J. Am. Chem. Soc., 2020, 142(14): 6516-6520. 33 [22] Zhong JH, Zhang J, Jin X, Liu JY, Li Q, Li MH, Cai W, Wu DY, Zhan D, 34 Ren B. Quantitative Correlation between Defect Density and 35 Heterogeneous Electron Transfer Rate of Single Layer Graphene[J]. J. Am. 36 Chem. Soc., 2014, 136(47): 16609-16617. 37 [23] Ferrari AC, Basko DM. Raman Spectroscopy as a Versatile Tool for 38 Studying the Properties of Graphene[J]. Nat. Nanotechnol., 2013, 8(4): 235-39 246 40 [24] Li W, Li Y, Xu K. Facile, Electrochemical Chlorination of Graphene from 41 an Aqueous Nacl Solution[J]. Nano Lett., 2021, 21(2): 1150-1155. 42 [25] Eckmann A, Felten A, Mishchenko A, Britnell L, Krupke R, Novoselov KS, 43 Casiraghi C. Probing the Nature of Defects in Graphene by Raman 44 Spectroscopy[J]. Nano Lett., 2012, 12(8): 3925-3930. 45 [26] Li B, Zhou L, Wu D, Peng HL, Yan K, Zhou Y, Liu ZF. Photochemical 46 Chlorination of Graphene[J]. Acs Nano, 2011, 5(7): 5957-5961. 47
- 48



GSI



2 Entry for the Table of Contents





)



1

单层石墨烯微米尺度图案化和功能化:调控电子传 2 输特性 3 4 崔苗苗a,韩联欢a,b*,曾兰平a,郭佳瑶a,宋维英a,刘川a,吴元菲a,罗 5 世翊a, 刘云华 c*, 詹东平 a* 6 (a. 化学化工学院化学系:固体表面物理化学国家重点实验室,电化学技术教育部工程研究中心,厦门大学, 7 厦门 361005 8 b厦门大学萨本栋微纳米科学与技术研究所机电工程系,厦门大学,厦门361005 9 c华中科技大学国家 CAD 支撑软件工程研究中心,武汉 430074) 10 11 摘要: 石墨烯具有优异的物理特性, 如单原子厚度、极高的载流子迁移率等。然而, 其零带隙的半金属特性限制了 12 13 题在高性能场效应晶体管中的应用。为此,研究者们提出了石墨烯纳米化、外场诱导、掺杂以及化学图案化等策略, 以调控其带隙宽度。但是,这些方法的可控性以及稳定性还需要进一步改善。在本研究中,我们提出采用电化学溴 14 化并结合光刻图案化调控单层石墨烯的电子传输特性。通过这种方法,我们成功制备了图案化的溴化石墨烯 15 16 (SLGBr)。进一步研究表明单层石墨烯的电子传输性能可以通过溴化程度来调控。当溴化程度较小时, SLGBr表 17 现为电阻特性,且其电导随溴化程度增加而减小;当溴化程度增加到一定值时,SLGBr表现为与半导体类似的特性。 本研究将为全石墨烯器件的制备提供可行的技术路线,拓展其在微电子领域的应用。 18 19 关键词:石墨烯图案化:电子传输;电化学溴化;光刻;全石墨烯器件 20