

Journal of Electrochemistry

Volume 20
Issue 6 Special Issue of Bioelectroanalytical
Chemistry (Editor: Professor XIA Xing-hua)

2014-12-28

Electrochemical Applications of Single-walled Carbon Nanohorns

Su-ping LI

Huai-min GUAN

Shu-yun ZHU

Muhammad Rehan Hassan Shah GILANI

Saima HANIF

Guo-bao XU

State Key Laboratory of Electroanalytical Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, China; guobaoxu@ciac.ac.cn

Yue-jin TONG

College of Chemistry and Chemical Engineering, Fujian Normal University, Fuzhou 350007, China;
tongyuejin@fjnu.edu.cn

Recommended Citation

Su-ping LI, Huai-min GUAN, Shu-yun ZHU, Muhammad Rehan Hassan Shah GILANI, Saima HANIF, Guo-bao XU, Yue-jin TONG. Electrochemical Applications of Single-walled Carbon Nanohorns[J]. *Journal of Electrochemistry*, 2014 , 20(6): 501-505.

DOI: 10.13208/j.electrochem.140433

Available at: <https://jelectrochem.xmu.edu.cn/journal/vol20/iss6/1>

This Review is brought to you for free and open access by Journal of Electrochemistry. It has been accepted for inclusion in Journal of Electrochemistry by an authorized editor of Journal of Electrochemistry.

单壁碳纳米角电化学应用

李素萍¹, 关怀民^{2,3}, 朱树芸⁴, GILANI Muhammad Rehan Hassan Shah⁴,
HANIF Saima⁴, 徐国宝^{1,4*}, 童跃进^{1,2*}

(1. 福建师范大学化学与化工学院,福建福州350007; 2. 福建师范大学化工新材料研究所,福建福州350007;
3. 福建师范大学材料科学与工程学院,福建福州350007;
4. 长春应用化学研究所电分析化学国家重点实验室,吉林长春1300224)

摘要: 单壁碳纳米角具有独特的性质,如大的比表面积、良好的导电性和生物相容性等,在某些领域应用广泛。本文结合作者课题组工作,综述了单壁碳纳米角电化学应用现状,并展望其今后的发展趋势。

关键词: 单壁碳纳米角;电化学;综述

中图分类号: O646

文献标识码: A

1999年Iijima研究组^[1]在氩气氛围下用激光烧蚀石墨法制得一端呈封闭的锥形(锥角约20°),而其余部分系与碳纳米管类似的石墨管状(管内径为2~5 nm,管长为30~50 nm)的单壁碳纳米角(SWCNHs),约2000根SWCNHs聚集于一起形成直径约80~100 nm大丽花状的SWCNHs聚集体,如图1所示。无金属催化剂就可制得SWCNHs,可避免金属杂质带来的干扰。其特殊的几何构型决定了其力学、化学和电子性质,如高比表面积、热稳定性、优良的导电性质,使其在吸附和存储气体^[2-3]、催化剂载体^[4]、药物载体^[5]、气体传感器^[6]、生物传感器^[7-8]和电化学储能^[9]等方面具有广泛的应用。

本文结合作者课题组工作,综述SWCNHs在电化学分析和电化学能源等的应用现状并展望其今后的发展趋势。

1 电化学应用

近年,SWCNHs主要在电催化、电化学分析、锂离子电池材料、电化学电容器以及燃料电池中金属纳米颗粒的催化载体等有重要应用。

1.1 SWCNHs 糊电极

2001年,碳糊电极首次应用于研究有机物质在电极上的反应机理^[10]。随着研究的深入,因碳糊

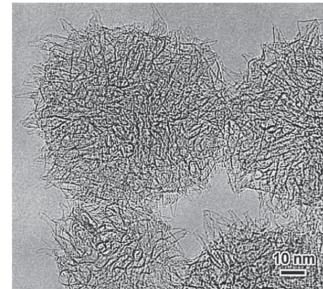


图 1 大丽花状 SWCNHs 聚集体的透射电镜照片^[1]

Fig. 1 TEM image of dahlia-like SWCNHs^[1]

电极制备简单等优点^[11],碳糊修饰电极已大量应用于电化学分析^[12]及生物传感器领域^[13]。碳纳米管(CNTs)的出现拓宽了碳糊电极的应用范围。同时CNTs糊电极显现了对抗坏血酸、尿酸、多巴胺、3,4-二羟基苯乙酸(DOPAC)及过氧化氢等有良好电催化活化作用^[14-17]。

为改善性能,作者课题组制作SWCNHs糊电极^[18],经活化处理提高检测H₂O₂的灵敏度和线性范围。与石墨糊电极及多壁CNTs电极相比,其线性范围更宽(0.5~100 mmol·L⁻¹)。SWCNHs的电化学活化可形成大量的缺陷,其边缘部分的含氧官能团更利于高浓度H₂O₂的检测。其大丽花状的

收稿日期: 2014-05-13, 修订日期: 2014-07-11 * 通讯作者, Tel: (86-431)85262747, E-mail: guobaoxu@ciac.ac.cn; tongyuejin@fjnu.edu.cn

国家自然科学基金项目(No. 21274022, No. 21175126)资助

聚集体提高了目标分析物接触的活性位点,呈现了良好的 H_2O_2 电化学性能,有望替代常用铂电极用于 H_2O_2 检测,特别是高浓度 H_2O_2 检测。SWCNHs 制备过程不使用金属催化剂,故 SWCNHs 不含金属杂质,对 H_2O_2 的电催化活性主要源于 SWCNHs。

1.2 SWCNHs 修饰电极

1) 固相萃取吸附剂

SWCNHs 作为固相萃取吸附剂,结合电化学法可快速、灵敏检测 4-硝基苯酚。与多壁碳纳米管(MWCNTs)修饰电极相比,该电极检测目标物 4-硝基苯酚的灵敏度提高 5.6 倍^[19]。

2) 苯二酚的异构体的检测

作者课题组应用 SWCNHs 修饰电极检测苯二酚的三种同分异构体(对苯二酚、儿茶酚和间苯二酚)^[20]。这三种苯二酚的同分异构体分布广泛,难以降解。邻、间、对苯二酚异构体的氧化电位在 SWCNHs 修饰电极上得以分离,并可实现同时检测。且在另两种苯二酚异构体存在下,其单一种苯二酚异构体的氧化电流仍随其浓度增加而线性增长。

3) 尿酸、多巴胺及抗坏血酸的检测

作者课题组研究表明,SWCNHs 修饰电极对尿酸(UA)、多巴胺(DA)和抗坏血酸(AA)有很好的电催化活性^[21]。三者扫描伏安曲线氧化还原电位在该电极上均明显负移,氧化还原电流峰增大,且可同时检测到。有意思的是,UA 在一般电极上的氧化是不可逆的,而在 SWCNHs 修饰电极上的氧化还原峰是可逆的。且 SWCNHs 修饰电极对 UA

具有很高的灵敏度,检测限较文献报道值低 1~2 个数量级(图 2)。

Zhu 等^[22]应用类似方法,快速直接定量地检测 L-色氨酸(Trp)和 L-酪氨酸(Tyr),SWCNHs 修饰电极对 Trp 和 Tyr 有很好的电催化氧化活性。Trp 和 Tyr 检测限分别达到 50 和 400 $\text{nmol}\cdot\text{L}^{-1}$ 。有望直接应用于血清实样的 Trp 和 Tyr 检测。Xu 等^[23]也应用 SWCNHs 修饰电极检测双酚 A。

1.3 电化学生物传感器

SWCNHs 溶解度系亟需解决的实际问题。文献曾报道 SWCNHs 功能化主要有共价修饰和非共价修饰法。这两种方法均可提高 SWCNHs 溶解度,以满足不同检测需要。作者课题组^[7]采用聚苯乙烯磺酸钠非共价修饰 SWCNHs。功能化 SWCNHs 能吸附肌红蛋白,用以构建检测 H_2O_2 电化学生物传感器,呈现出良好的电催化活性。响应线性范围和检测限分别为 3~350 $\mu\text{mol}\cdot\text{L}^{-1}$ 和 0.5 $\mu\text{mol}\cdot\text{L}^{-1}$ 。此外,Yang 等^[24]应用 SWCNHs 纳米材料信号标记三明治型双酶生物催化构建电化学免疫传感器,提高癌症生物标志物的检测灵敏度。Zhang 等^[25]提出快速检测微囊藻毒素-LR 的电化学免疫传感器。Liu 等^[26]还应用 SWCNHs-空心 Pt 纳米球信号标记构建了检测降钙素原的新型电化学免疫传感器。Qian 组^[27]和 Dai 组^[28]等亦报道了相关的研究。SWCNHs 的高比表面积及良好的电催化性能使得该类传感器表现出对目标分析物良好的灵敏度。

作者课题组还首次构建了 SWCNHs 的葡萄糖电化学生物传感器^[29]。将葡萄糖氧化酶固定于

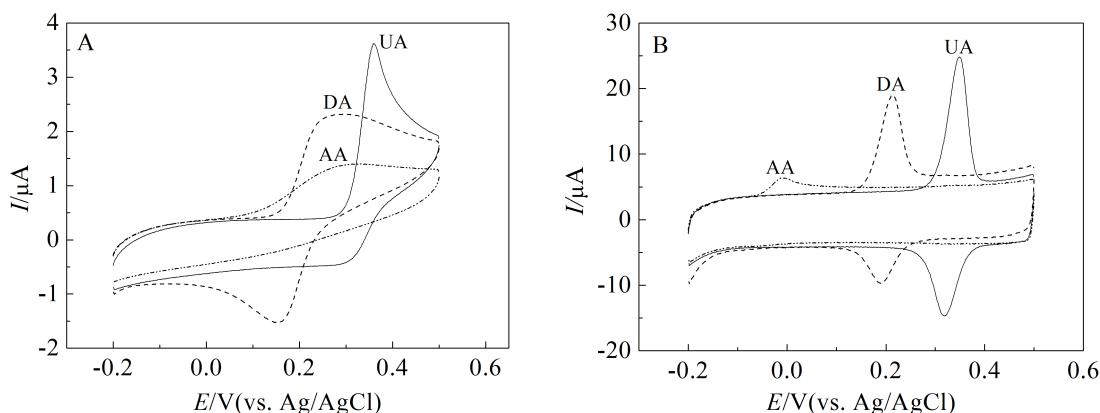


图 2 玻碳电极(A)和 SWCNHs 修饰电极(B)在 0.1 mol·L⁻¹ 磷酸缓冲 pH 7.0 溶液中尿酸(UA, 100 $\mu\text{mol}\cdot\text{L}^{-1}$)、多巴胺(DA, 100 $\mu\text{mol}\cdot\text{L}^{-1}$)和抗坏血酸(AA, 100 $\mu\text{mol}\cdot\text{L}^{-1}$)的循环伏安曲线, 扫速 50 mV·s⁻¹^[21]

Fig. 2 CVs of uric acid (UA, 100 $\mu\text{mol}\cdot\text{L}^{-1}$), dopamine (DA, 100 $\mu\text{mol}\cdot\text{L}^{-1}$), and ascorbic acid (AA, 100 $\mu\text{mol}\cdot\text{L}^{-1}$) at bare GCE (A) and SWCNHs-modified GCE (B) in 0.1 mol·L⁻¹ pH 7.0 PBS, scan rate: 50 mV·s⁻¹^[21]

Nafion-SWCNHs 复合材料中,在缓冲溶液条件下,其呈现良好的电催化活性。另外,还可将大豆过氧化物酶固定于 SWCNHs, 实现大豆过氧化物酶的直接电子转移, 构建了无媒介体的 H₂O₂ 生物传感器^[8]。

1.4 新能源电池

Xu 等^[30]构建纳米介孔锐钛矿 TiO₂/SWCNHs 复合材料, 应用于锂离子电池。Zhao 等^[31]通过简单水热合成锂离子电池 Fe₂O₃/SWCNHs 复合材料, 该材料在高电流密度下显示出良好的循环稳定性。另外, 他们还合成了锂离子电池 SnO₂/SWCNHs 复合材料^[32]。Aissa 等^[33]应用 SWCNHs 包覆微纤维构筑电化学电源独立电极。

SWCNHs 还可应用于燃料电池中的电催化载体^[34-37]或基底材料^[38], 用于增强聚合物电解质膜燃料电池的性能。SWCNHs 作为金属纳米颗粒催化载体^[39], 较传统炭黑性能更佳。Dong 组应用 SWCNHs 作为氧化还原中介及电催化载体构建生物燃料电池^[40], 该葡萄糖/O₂ 生物燃料电池在生理条件下具有良好的性能。

1.5 电化学电容器

Yuge 等构建了电化学电容电极^[41-42], 该电极的比电容达到 100 F·g⁻¹。Izadi-Najafabadi 等^[43]合成 SWCNHs/SWCNTs(80%/20%, by mass) 复合材料用于高功率超级电容器电极。该新型复合电极具有较高的中-大孔体积, 可容纳更多电解液, 确保离子传输, 从而实现更高的功率。

2 展望

单壁碳纳米角特殊的结构决定了该材料具有独特的性能, 使其广泛应用于电化学的诸多领域如电化学分析、生物传感器、锂离子电池、燃料电池和太阳能电池^[44]等。但单壁碳纳米角在电化学领域的应用仍处于起步阶段, 今后还需不断地探索和拓宽。

SWCNHs 在电化学领域的研究和应用可能以下几个方面展开: ①开发 SWCNHs 自身的催化性能, 拓宽其在电催化和电化学分析的应用范围; ②发挥 SWCNHs 吸附能力强的优点, 通过富集提高分析的灵敏度或作为新型载体; ③制备各种 SWCNHs 复合材料, 开发 SWCNHs 复合材料电化学新应用。

参考文献 (References):

- [1] Iijima S, Yudasaka M, Yamada R, et al. Nano-aggregates of single-walled graphitic carbon nano-horns[J]. Chemical Physics Letters, 1999, 309(3/4): 165-170.
- [2] Liu Y, Brown C M, Neumann D A, et al. Metal-assisted hydrogen storage on Pt-decorated single-walled carbon nanohorns[J]. Carbon, 2012, 50(13): 4953-4964.
- [3] Krungleviciute V, Ziegler C A, Banjara S R, et al. Neon and CO₂ adsorption on open carbon nanohorns[J]. Langmuir, 2013, 29(30): 9388-9397.
- [4] Kosaka M, Kuroshima S, Kobayashi K, et al. Single-wall carbon nanohorns supporting Pt catalyst in direct methanol fuel cells[J]. The Journal of Physical Chemistry C, 2009, 113(20): 8660-8667.
- [5] Ajima K, Yudasaka M, Murakami T, et al. Carbon nanohorns as anticancer drug carriers[J]. Molecular Pharmaceutics, 2005, 2(6): 475-480.
- [6] Penza M, Aversa P, Cassano G, et al. Layered SAW gas sensor with single-walled carbon nanotube-based nanocomposite coating[J]. Sensors and actuators B: Chemical, 2007, 127(1): 168-178.
- [7] Liu X, Li H, Wang F, et al. Functionalized single-walled carbon nanohorns for electrochemical biosensing[J]. Biosensors & Bioelectronics, 2010, 25(10): 2194-2199.
- [8] Shi L, Liu X, Niu W, et al. Hydrogen peroxide biosensor based on direct electrochemistry of soybean peroxidase immobilized on single-walled carbon nanohorn modified electrode[J]. Biosensors & Bioelectronics, 2009, 24(5): 1159-1163.
- [9] Yang C, Kim Y, Endo M, et al. Nanowindow-regulated specific capacitance of supercapacitor electrodes of single-wall carbon nanohorns[J]. Journal of the American Chemical Society, 2007, 129(1): 20-21.
- [10] Švancara I, Vytrás K, Barek J, et al. Carbon paste electrodes in modern electroanalysis[J]. Critical Reviews in Analytical Chemistry, 2001, 31(4): 311-345.
- [11] Zakharchuk N F, Meyer B, Henning H, et al. A comparative study of Prussian-Blue-modified graphite paste electrodes and solid graphite electrodes with mechanically immobilized Prussian Blue[J]. Journal of Electroanalytical Chemistry, 1995, 398(1): 23-35.
- [12] Kalcher K, Kauffmann J M, Wang J, et al. Sensors based on carbon paste in electrochemical analysis: A review with particular emphasis on the period 1990-1993 [J]. Electroanalysis, 1995, 7(1): 5-22.
- [13] Matuszewski W, Trojanowicz M. Graphite paste-based enzymatic glucose electrode for flow injection analysis [J]. Analyst, 1988, 113(5): 735-738.
- [14] Rubianes M D, Rivas G A. Carbon nanotubes paste elec-

- trode[J]. *Electrochemistry Communications*, 2003, 5(8): 689-694.
- [15] Antiochia R, Gorton L. Development of a carbon nanotube paste electrode osmium polymer-mediated biosensor for determination of glucose in alcoholic beverages[J]. *Biosensors and Bioelectronics*, 2007, 22(11): 2611-2617.
- [16] Sanghavi B J, Srivastava A K. Simultaneous voltammetric determination of acetaminophen, aspirin and caffeine using an *in situ* surfactant-modified multiwalled carbon nanotube paste electrode[J]. *Electrochimica Acta*, 2010, 55(28): 8638-8648.
- [17] Ensaifi A, Karimi-Maleh H, Mallakpour S. A new strategy for the selective determination of glutathione in the presence of nicotinamide adenine dinucleotide (NADH) using a novel modified carbon nanotube paste electrode[J]. *Colloids and Surfaces B: Biointerfaces*, 2013, 104: 186-193.
- [18] Zhu S, Fan L, Liu X, et al. Determination of concentrated hydrogen peroxide at single-walled carbon nanohorn paste electrode [J]. *Electrochemistry Communications*, 2008, 10(5): 695-698.
- [19] Zhu S, Niu W, Li H, et al. Single-walled carbon nanohorn as new solid-phase extraction adsorbent for determination of 4-nitrophenol in water sample[J]. *Talanta*, 2009, 79(5): 1441-1445.
- [20] Zhu S, Gao W, Zhang L, et al. Simultaneous voltammetric determination of dihydroxybenzene isomers at single-walled carbon nanohorn modified glassy carbon electrode[J]. *Sensors and Actuators B: Chemical*, 2014, 198: 388-394.
- [21] Zhu S, Li H, Niu W, et al. Simultaneous electrochemical determination of uric acid, dopamine, and ascorbic acid at single-walled carbon nanohorn modified glassy carbon electrode[J]. *Biosensors & Bioelectronics*, 2009, 25(4): 940-943.
- [22] Zhu S, Zhang J, Zhao X E, et al. Electrochemical behavior and voltammetric determination of L-tryptophan and L-tyrosine using a glassy carbon electrode modified with single-walled carbon nanohorns[J]. *Microchimica Acta*, 2014, 181(3/4): 445-451.
- [23] Xu G, Gong L, Dai H, et al. Electrochemical bisphenol A sensor based on carbon nanohorns[J]. *Analytical Methods*, 2013, 5(13): 3328-3333.
- [24] Yang F, Han J, Zhuo Y, et al. Highly sensitive impedimetric immunosensor based on single-walled carbon nanohorns as labels and bienzyme biocatalyzed precipitation as enhancer for cancer biomarker detection[J]. *Biosensors & Bioelectronics*, 2014, 55: 360-365.
- [25] Zhang J, Lei J, Xu C, et al. Carbon nanohorn sensitized electrochemical immunosensor for rapid detection of microcystin-LR[J]. *Analytical Chemistry*, 2010, 82(3): 1117-1122.
- [26] Liu F, Xiang G, Chen X, et al. A novel strategy of procalcitonin detection based on multi-nanomaterials of single-walled carbon nanohorns-hollow Pt nanospheres/PAMAM as signal tags[J]. *RSC Advances*, 2014, 4(27): 13934-13940.
- [27] Qian R, Ding L, Bao L, et al. *In situ* electrochemical assay of cell surface sialic acids featuring highly efficient chemoselective recognition and a dual-functionalized nanohorn probe[J]. *Chemical Communications*, 2012, 48 (32): 3848-3850.
- [28] Dai H, Yang C, Ma X, et al. A highly sensitive and selective sensing ECL platform for naringin based on beta-cyclodextrin functionalized carbon nanohorns[J]. *Chemical Communications*, 2011, 47(43): 11915-11917.
- [29] Liu X, Shi L, Niu W, et al. Amperometric glucose biosensor based on single-walled carbon nanohorns[J]. *Biosensors & Bioelectronics*, 2008, 23(12): 1887-1890.
- [30] Xu W, Wang Z, Guo Z, et al. Nanoporous anatase TiO₂/single-wall carbon nanohorns composite as superior anode for lithium ion batteries[J]. *Journal of Power Sources*, 2013, 232: 193-198.
- [31] Zhao Y, Li J, Ding Y, et al. Single-walled carbon nanohorns coated with Fe₂O₃ as a superior anode material for lithium ion batteries[J]. *Chemical Communications*, 2011, 47(26): 7416-7418.
- [32] Zhao Y, Li J, Ding Y, et al. A nanocomposite of SnO₂ and single-walled carbon nanohorns as a long life and high capacity anode material for lithium ion batteries[J]. *RSC Advances*, 2011, 1(5): 852-856.
- [33] Aissa B, Hamoudi Z, Takahashi H, et al. Carbon nanohorns-coated microfibers for use as free-standing electrodes for electrochemical power sources[J]. *Electrochemistry Communications*, 2009, 11(4): 862-866.
- [34] Gattia D M, Antisari M V, Giorgi L, et al. Study of different nanostructured carbon supports for fuel cell catalysts [J]. *Journal of Power Sources*, 2009, 194(1): 243-251.
- [35] Boaventura M, Brandao L, Mendes A. Single-wall nanohorns as electrocatalyst support for high temperature PEM fuel cells[J]. *Journal of the Electrochemical Society*, 2011, 158(4): B394-B401.
- [36] Brandao L, Boaventura M, Ribeirinha P. Single wall nanohorns as electrocatalyst support for vapour phase high temperature DMFC[J]. *International Journal of Hydrogen Energy*, 2012, 37(24): 19073-19081.

- [37] Niu B, Xu W, Guo Z, et al. Controllable deposition of platinum nanoparticles on single-wall carbon nanohorns as catalyst for direct methanol fuel cells[J]. *Journal of Nanoscience and Nanotechnology*, 2012, 12(9): 7376-7381.
- [38] Yuan D, Zeng J, Chen J, et al. Synthesis of hollow-cone-like carbon and its application as support material for fuel cells[J]. *Journal of the Electrochemical Society*, 2009, 156(3): B377-B380.
- [39] Brandao L, Boaventura M, Passeira C, et al. An electrochemical impedance spectroscopy study of polymer electrolyte membrane fuel cells electrocatalyst single wall carbon nanohorns-supported[J]. *Journal of Nanoscience and Nanotechnology*, 2011, 11(10): 9016-9024.
- [40] Wen D, Deng L, Zhou M, et al. A biofuel cell with a single-walled carbon nanohorn-based bioanode operating at physiological condition[J]. *Biosensors & Bioelectronics*, 2010, 25(6): 1544-1547.
- [41] Wang Z, Luan D, Madhavi S, et al. Assembling carbon-coated alpha-Fe₂O₃ hollow nanohorns on the CNT backbone for superior lithium storage capability[J]. *Energy & Environmental Science*, 2012, 5(1): 5252-5256.
- [42] Yuge R, Manako T, Nakahara K, et al. The production of an electrochemical capacitor electrode using holey single-wall carbon nanohorns with high specific surface area [J]. *Carbon*, 2012, 50(15): 5569-5573.
- [43] Izadi-Najafabadi A, Yamada T, Futaba D N, et al. High-power supercapacitor electrodes from single-walled carbon nanohorn/nanotube composite[J]. *ACS Nano*, 2011, 5(2): 811-819.
- [44] Casillas R, Lodermeier F, Costa R D, et al. Substituting TiCl₄-carbon nanohorn interfaces for dye-sensitized solar cells[J]. *Advanced Energy Materials*, 2014, 4(6), DOI: 10.1002/aenm.201301577.

Electrochemical Applications of Single-walled Carbon Nanohorns

LI Su-ping¹, GUAN Huai-min^{2,3}, ZHU Shu-yun⁴, GILANI Muhammad Rehan Hassan Shah⁴, HANIF Saima⁴, XU Guo-bao^{1,4*}, TONG Yue-jin^{1,2*}

(1. College of Chemistry and Chemical Engineering, Fujian Normal University, Fuzhou 350007, China;

2. Institute of New Chemical Materials, Fujian Normal University, Fuzhou 350007, China;

3. College of Materials Science and Engineering, Fujian Normal University, Fuzhou 350007, China;

4. State Key Laboratory of Electroanalytical Chemistry, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, China)

Abstract: Single-wall carbon nanohorns (SWCNHs) have unique properties, such as a large specific surface area, good electrical conductivity and biocompatibility. It has been widely utilized in many fields. In the present review, the progress in electrochemistry study of SWCNHs electrochemistry study has been summarized and the future research trends have been proposed.

Key words: single-walled carbon nanohorns; electrochemistry; review