

Journal of Electrochemistry

Volume 17 | Issue 2

2011-05-28

Capacitance Behaviors of Ti/Ru_{0.1}Ti_{0.1}Sn_{0.802} Electrodes

Hai-Yan CHEN

Yan-Qun SHAO

Dian TANG

Recommended Citation

Hai-Yan CHEN, Yan-Qun SHAO, Dian TANG. Capacitance Behaviors of Ti/Ru_{0.1}Ti_{0.1}Sn_{0.802} Electrodes[J]. *Journal of Electrochemistry*, 2011 , 17(2): Article 20.

DOI: 10.61558/2993-074X.2835

Available at: <https://jelectrochem.xmu.edu.cn/journal/vol17/iss2/20>

This Research Notes is brought to you for free and open access by Journal of Electrochemistry. It has been accepted for inclusion in Journal of Electrochemistry by an authorized editor of Journal of Electrochemistry.

文章编号:1006-3471(2011)02-0231-03

Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 电极电容性能研究

陈海燕, 邵艳群*, 唐 电

(福州大学材料研究所, 福建 福州 350108)

摘要: 低温热分解法制备 Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂. XRD 表征该样品结构特性, 循环伏安和恒流充放电测定电极性能. 结果表明, Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 物相属细小的金红石相, 在 20 mV/s 扫速下该电极比电容达 933 F/g. 1000 次充放电循环后, 比电容衰减 23.6%, 显示良好的循环稳定性和可逆性.

关键词: 超级电容器; RuO₂-TiO₂-SnO₂; 热分解法; 比电容

中图分类号: TM911; TQ174

文献标识码: A

RuO₂ 电极超级电容器比电容可达 185 F/g^[1], 但 Ru 价格昂贵, 限制其实际应用. 可替代 RuO₂ 的其它氧化物大多是过渡金属氧化物, 如 MnO₂^[2]、Co₃O₄^[3]、NiO^[4]、TiO₂^[5]、SnO₂^[6] 等. 研究表明, 在 RuO₂ 中掺杂第 2 组分也可提高电极比电容^[7-8]. 本文应用低温热分解法制备 RuO₂-TiO₂-SnO₂ 氧化物涂层, 分析电极的物相结构并测定电极电容性能. 结果显示在扫描速率为 20 mV/s 下, Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 电极的比电容达到 933 F/g.

1 测试与体系

将 RuCl₃ (36.9%)、C₁₆H₃₆O₄Ti (98%) 和 SnCl₄ (98%) 按 1:1:8 (by mass) 比例分别溶于适量无水乙醇, 超声振荡均匀分散, 搁置 12 h, 取适量混合液均匀涂覆于钛板上, 红外干燥, 箱式炉中 300 °C 下氧化 10 min, 出炉、冷却. 如此多次重复涂覆, 最后将钛基涂层烘干、退火 (300 °C, 1 h) 空冷.

使用 Philips Xpert-MPD 衍射仪分析涂层结构, CuK α_1 辐射, 管电压 40 kV, 电流 40 mA. CHI660C 电化学工作站作循环伏安测试和恒流充放电测试, 工作电极 Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ (1 cm²), 参比电极 232 型饱和甘汞电极, 钛板作辅助电极, 电解液 0.5 mol/L H₂SO₄.

2 结果与讨论

2.1 XRD 谱图

图 1 是 Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 电极的 XRD 图谱,

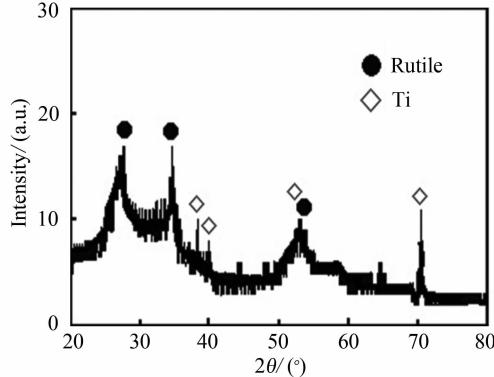


图 1 Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 电极 XRD 图谱

Fig. 1 XRD pattern of Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ electrode

显示该样品由金红石相和 Ti 组成. 图中, 金红石相的 3 个衍射峰明显宽化且不完全对称, 表明该涂层的晶粒细小并存在相分离. 这与 SnO₂ 易形成较多细小金红石晶粒有关^[9]. 但如 SnO₂ 含量较多, 则仅部分能与 RuO₂ 形成固溶体^[7,9].

2.2 循环伏安曲线

图 2 示出 Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 电极在不同扫速下的循环伏安曲线. 如图, 伏安曲线呈现出较好的矩形特征, 即电极可逆性较好. 扫速增大, 氧化还原峰电位差增大, 同时电极比电容逐减 (见表 1).

2.3 恒电流充放电曲线

图 3 绘出 Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ 电极在 5 mA/

表 1 不同扫速下 $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ 电极的比电容Tab. 1 Specific capacitance of the $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ electrode at different scan rates

Scan rate/ $\text{mV} \cdot \text{s}^{-1}$	5	10	15	20	25	35
$C_s, \text{RuO}_2/\text{F} \cdot \text{g}^{-1} \cdot \text{cm}^{-2}$	1022	979	960	933	883	830

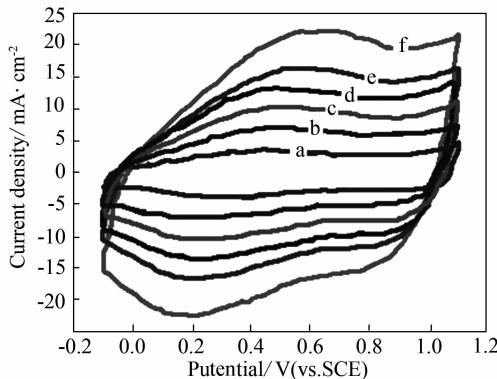
图 2 不同扫速 $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ 电极的循环伏安曲线

Fig. 2 Cyclic voltammograms of $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ electrode at different scan rates
a ~ f/ $\text{mV} \cdot \text{s}^{-1}$: 5; 10; 15; 20; 25; 35

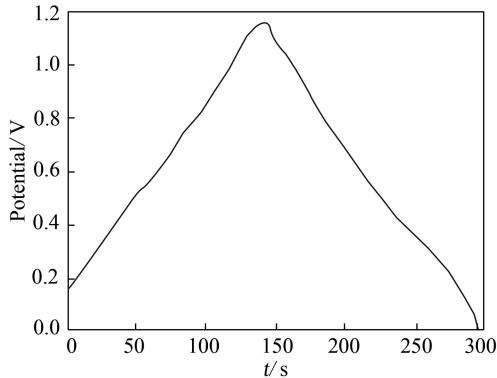
图 3 $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ 电极于 $5 \text{ mA}/\text{cm}^2$ 的恒电流充放电曲线

Fig. 3 Charge/discharge curves of $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ electrode at current density of $5 \text{ mA}/\text{cm}^2$

cm^2 恒电流下充放电曲线. 由图看出, 电极充放电电压随时间大体呈线性变化, 且接近对称, 显示出较好的电极可逆性和较好的电容特性.

2.4 电极稳定性

图 4 是 $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ 电极在扫速 $20 \text{ mV}/\text{s}$ 、电位窗口为 $-0.1 \sim 1.1 \text{ V}$ 下的比电容循环寿命曲线. 由图可见, 电极比容量达 $933 \text{ F} \cdot \text{g}^{-1} \cdot \text{cm}^{-2}$.

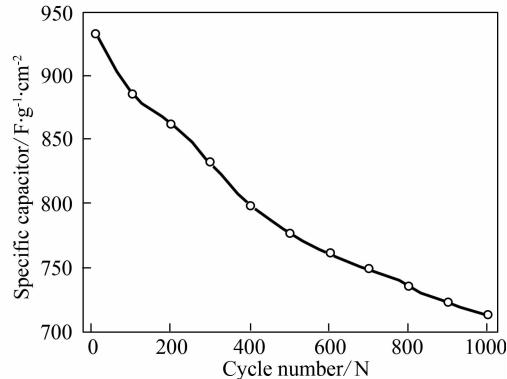
图 4 扫速 $20 \text{ mV}/\text{s}$ 电位窗口 $-0.1 \sim 1.1 \text{ V}$ 时 $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ 电极比电容循环寿命

Fig. 4 Cycle life of $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ electrode at potential window of $-0.1 \sim 1.1 \text{ V}$ and scan rate $20 \text{ mV} \cdot \text{s}^{-1}$

1000 次循环后, 比电容衰减 23.6%.

3 结 论

由低温热分解法制备的仅含 10% RuO_2 的 $\text{Ti}/\text{Ru}_{0.1}\text{Ti}_{0.1}\text{Sn}_{0.8}\text{O}_2$ 涂层, 添加 Sn 组元可使晶粒细化, 导致该样品特征衍射峰宽化. 在 $20 \text{ mV}/\text{s}$ 扫描速率下, 该电极比电容达 933 F/g , 1000 次充放电循环后, 电极比电容衰减 23.6%.

参考文献(References):

- [1] Lin K M, Chang K H, Hu C C, et al. Mesoporous RuO_2 for the next generation supercapacitors with an ultrahigh power density [J]. *Electrochimica Acta*, 2009, 54 (19): 4574-4581.
- [2] Wen Jian-guo(文建国), Ruan Xiang-yuan(阮湘元), Zhou Zhen-tao(周震涛). Study on preparation and electrochemical pseudocapacitance characteristics of $\text{Ru}_{0.1}\text{Mn}_{0.9}\text{O}_x$ [J]. *Rare Metal Materials And Engineering(稀有金属材料与工程)*, 2009, 38(5): 931-934.
- [3] Yuan An-bao(袁安保), Zhang Qing-lin(章庆林). Solid reaction preparation and electrochemical characteristics of nano-structured Co_3O_4 material for supercapacitor [J]. *Journal of Functional Materials and Devices(功能材料与*

- 器件学报),2007,13(1):1-6.
- [4] Chen Ye(陈野),Liu Liang(刘良),Zhang Zun-bo(张尊波),et al. Preparation of NiO by parallel flow precipitation process and its capacitance performance [J]. Fine Chemicals(精细化工),2008,25(5):424-427.
- [5] Chang K H,Hu C C. Hydrothermal synthesis of binary Ru-Ti oxides with excellent performances for supercapacitors [J]. Electrochimica Acta,2006,52(4):1749-1757.
- [6] Hu C C,Chang K H,Wang C C. Two-step hydrothermal synthesis of Ru-Sn oxide composites for electrochemical supercapacitors[J]. Electrochimica Acta,2007,52(13):4411-4418.
- [7] Wang Xin,Tang Dian,Zhou Jin-gen. Microstructure,morphology and electrochemical property of RuO₂70SnO₂30 mol% and RuO₂30SnO₂70mol% coatings[J]. Alloys and Compounds,2007,430(1/2):60-66.
- [8] Wang Ling-li(王玲利),Peng Qiao(彭乔). Advances in research on Ti anodes coating with RuO₂ [J]. Liaoning Chemical Industry(辽宁化工),2006,35(8):485-487.
- [9] Chen Yong-yi(陈永毅),Wang Xin(王欣),Shao Yan-qun(邵艳群),et al. Structure and morphology of titanium anode coating with different combinations from Ru-Ti and Ir-Ta [J]. The Chinese Journal of Nonferrous Metals(中国有色金属学报),2009,19(4):689-694.

Capacitance Behaviors of Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ Electrodes

CHEN Hai-yan, SHAO Yan-qun*, TANG Dian

(Institute for Materials Research, Fuzhou University, Fuzhou 350108, China)

Abstract: The Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ electrodes were prepared by low temperature thermal decomposition. The structural characteristics were analyzed by X-ray diffraction(XRD). The performances were tested by cyclic voltammetry and constant current charge-discharge measurements. The results show that, the Ti/Ru_{0.1}Ti_{0.1}Sn_{0.8}O₂ belongs to small rutile phase. A high specific capacitance of 933 F/g is obtained at scan rate of 20 mV/s. The cycle life test shows a 23.6% specific capacitance lost after 1000 cycles. The electrode has a good cycle stability and reversibility.

Key words: supercapacitor; RuO₂-TiO₂-SnO₂; thermal decomposition; specific capacitance